732/Chem

UG/6th Sem/CHEM-H-DSE-T-3/22

U.G. 6th Semester Examination - 2022 CHEMISTRY

[HONOURS]

Discipline Specific Elective (DSE)
Course Code: CHEM-H-DSE-T-3
(Advanced Physical Chemistry)

Full Marks : 40 Time : $2\frac{1}{2}$ Hours

The figures in the right-hand margin indicate marks.

Candidates are required to give their answers in their own words as far as practicable.

1. Answer any **five** questions:

 $2 \times 5 = 10$

- i) Define thermodynamic probability.
- ii) Write the applications of Boltzmann's distribution law.
- iii) Calculate the angle at which a first order reflection which occurs in a x-ray spectrometer when x-rays of wavelength $1.52\ A^0$ are diffracted by atoms of a crystal. Given that inter-planar distance is $3.8\ A^0$.
- iv) Justify Dulong-Petit's law from classical mechanics.

[Turn Over]

- v) Write the difference between thermoplastic and thermosetting polymers.
- vi) Calculate the inter-planar spacing ratio d_{100} : d_{110} : d_{111} for FCC lattice.
- vii) Draw the planes with Miller Indices (101) and (111).
- 2. Answer any two questions from the following:

 $5 \times 2 = 10$

- i) Express entropy, enthalpy, free energy, internal energy and heat capacity at constant volume in terms of molar partition function.
- ii) a) Deduce the relation between entropy and thermodynamic probabilit.
 - b) If the radius of an element is 400 pm and it crystallizes as FCC lattice, calculate the edge length of unit cell. 3+2=5
- iii) Derive Einstein's equation for heat capacity of solids and write its limitations. 3+2=5
- 3. Answer any two questions from the following:

 $10 \times 2 = 20$

i) a) Derive Boltzmann distribution equation with evaluation of β -value.

732/Chem.

(2)

- b) Describe Osmometric method for the determination of molecular weight of polymers.
- ii) a) Explain the kinetics of free radical mechanism of addition polymerization .
 - b) Write the third law of thermodynamics and explain how absolute entropy value at any temperature can be determined using third law.
 - A compound having BCC geometry has molar mass 50. Calculate the density of the unit cell if its edge length is 290 pm.

$$3+(1+3)+3=10$$

- iii) a) Deduce Bragg's Equation for crystal study and write its significance.
 - b) Distinguish between tetragonal and orthorhombic crystals with examples.
 - c) Using Boltzmann's molecular distribution equation, calculate the ratio of populations at 25°C of energy levels separated by 400 KJ of energy per mole assuming the energy levels to be non-degenerate.

732/Chem.

(3)